



RECENT DEVELOPMENTS IN ION MOLECULE REACTION – MASS SPECTROMETRY FOR CO₂ QUALITY MONITORING

Front Door to Back Door

BevTech'16, Ft. Myers

Dr. Siegfried Praun, Arno Weissnicht, Dr. Christian Leidlmair, Hannes Zingerle & Alexander Genuin M.Sc. (V&F Analyse- und Messtechnik GmbH)

And

Mark Taylor, Nicole James & Dr. Don Pachuta (Airborne Labs International, Inc)

- **IMR – MS Technology using a *Dual Capillary Inlet System***
- **Latest Applications**
 - Final Product ISBT Purity Gas Monitoring
 - Catox Optimization
 - Fermentation Plant Feed- In-Process – Final Product
 - Natural Well Feed Gas Profiles & GC Correlation
 - TSC Reductive Converter *Extended* Feed Gas Applications

Beverages / Foods / Packaging

- CO₂
- Beverages
- Food
- Flavors
- Flavors: good/bad breath + nose-space
- Mold in beverages, jelly
- PET Bottle Contamination
Monitoring



IMR-MS Technology

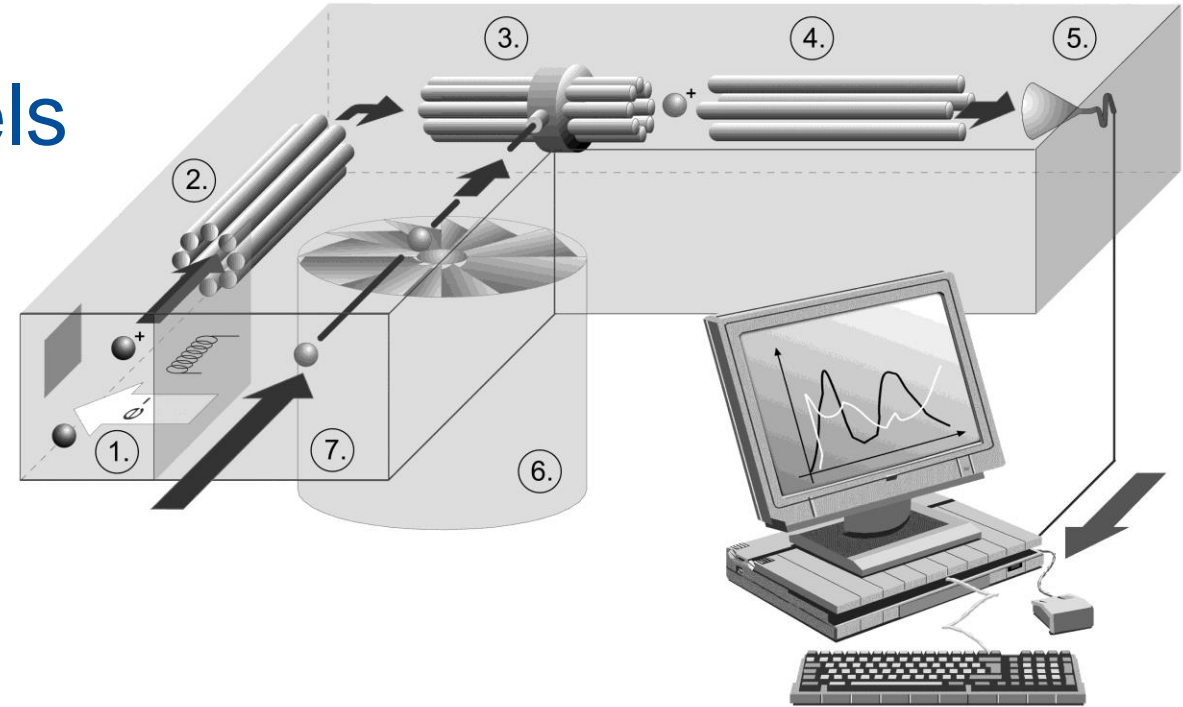
Use of discrete energy levels

LE: 10.44 eV

ME: 12.13 eV

HE: 13.99 eV

followed by



1. Primary Ion Source
2. Octopole Separation Device

3. Charge Exchange Cell
4. Quadrupole - Mass Filter
5. Particle Detector

6. Vacuum System
7. Gas Inlet System

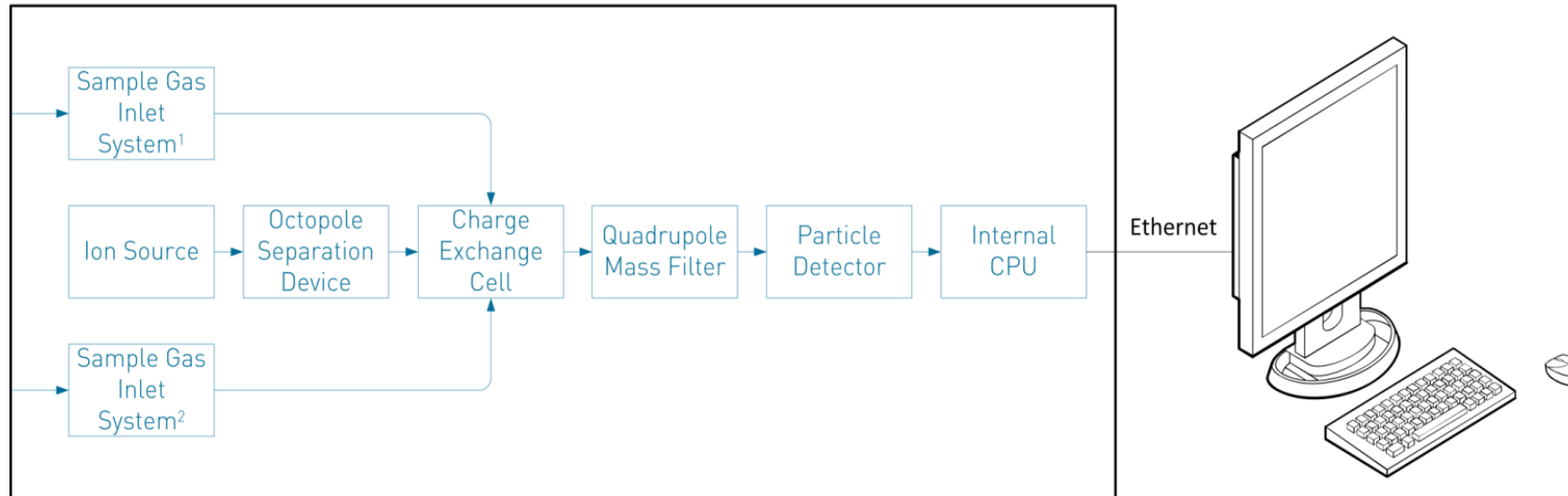
Ion Molecule Reaction with sample gas

molecules & Mass Spectrometry = IMR - MS



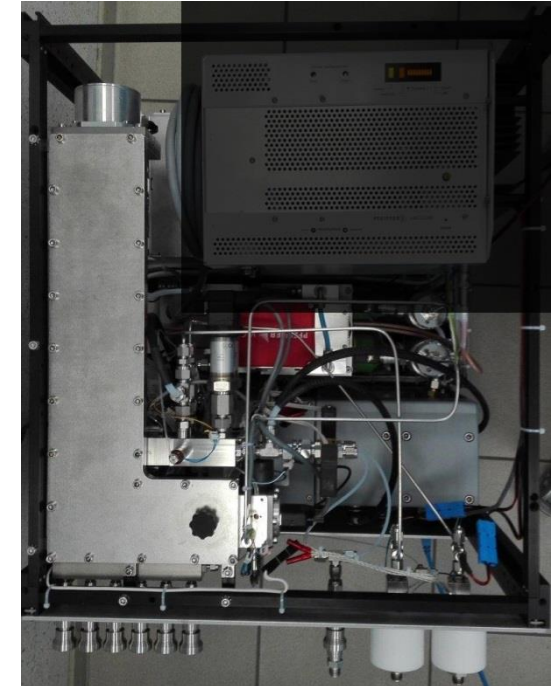
IMR-MS Technology – Dual Capillary Inlet

Schematic Dual Sample Gas Inlet - IMR - MS



¹ Sample gas inlet for final CO₂ product quality

² Optional sample gas inlet for CO₂ feed- and process gas analysis



IMR-MS Technology

- Single two-body reactive collision conditions in charge exchange chamber
 - Neutral, purely kinetic collisions → no change of ion mass number
 - Very low operating pressure → long mean free path between particles → Avoidance of bulk gas ionization
- Simultaneous detection of organic and inorganic compounds resp. compound classes (online or offline)
- Concentrations: ppt – ppb - ppm - Vol%
- Time of analysis: > 1 msec per compound

IMR-MS Technology - CO₂ Applications

Feed Gas

- Sources: combustion, natural wells, fermentation, geothermal emission, ammonia, coal gasification, ethylene oxide, phosphate rocks, etc.
- CO₂ Impurities: ppb – Vol %
- Online monitoring of impurity profiles & time variations



In-Process

- Online monitoring of CO₂ purification steps of catox, hydrocat, hydrolyzer, carbon beds, scrubbers, filters, etc.



Final Product Quality Control

- Product to Storage, Storage Tanks, Truck filling stations



- **CO₂ Feed Gas - Petrochemical Plant – Background:**
 - Catox malfunction in contrast to its design specs
 - Time-Related decrease in Catox conversion rate
 - Regeneration of Catox activity at lower temperatures
 - Breakthrough of small organics in the CO₂ feed gas
 - IMR-MS Monitoring of major impurities: O₂, H₂O, methane, ethylene, propylene, acetaldehyde, ethylene oxide, etc.

A. Original CATOX: 0.3% Pt on Al₂O₃

- T-in Catox from 390 – 360 C: 1.5 – 7.5 ppm ethylene (maximum desired value: 1 ppm)
- Addition of adsorbent → no better performance
- Possible solution: higher operating Catox temperature
- Catalytic oxidation: exothermal → too high temperature → high risk for damage of Catox

B. New Catox: 0.15% Pt + 0.15% Pd on Al₂O₃

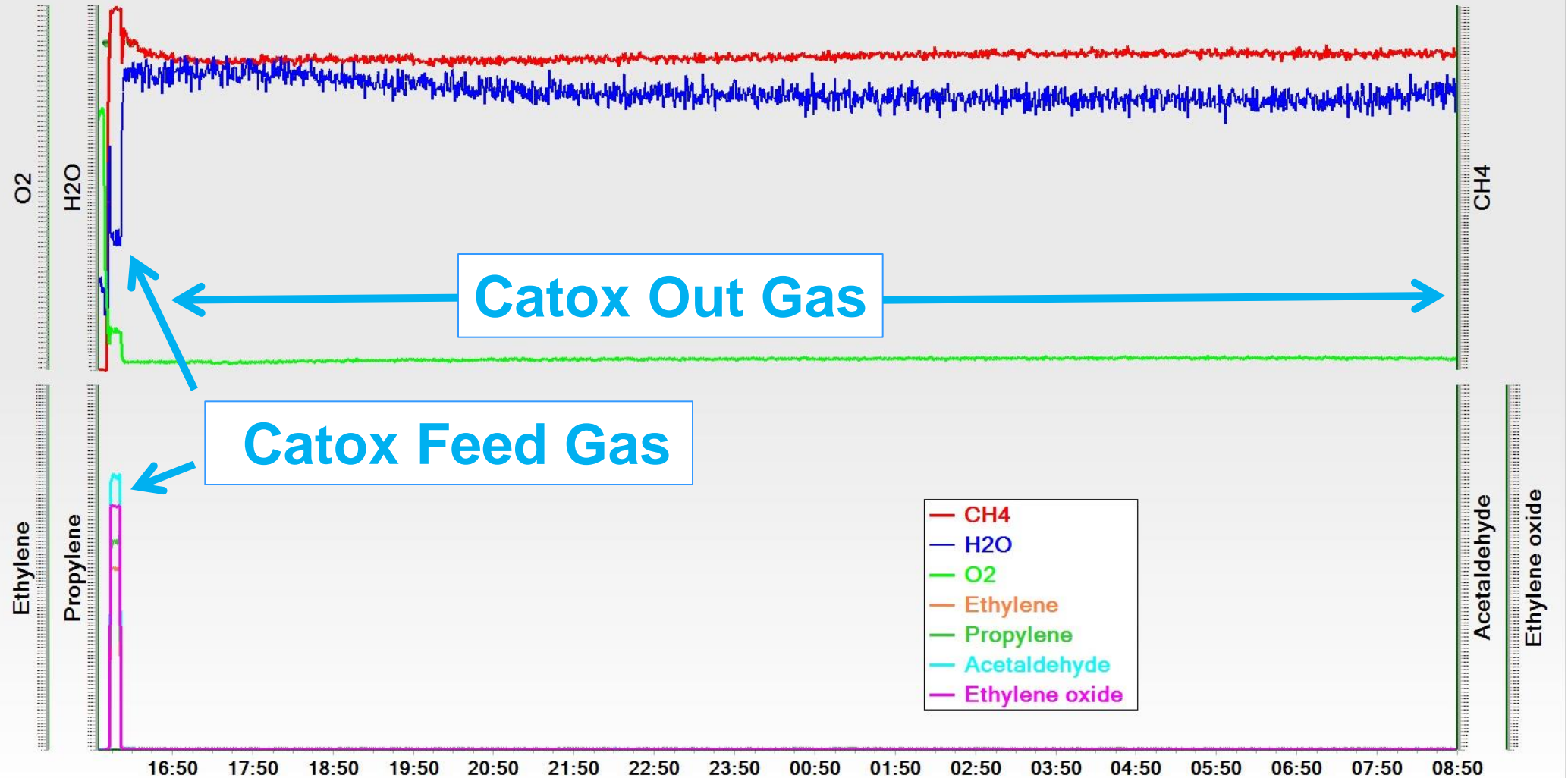
- At T-in 360 – 380 C: worse performance than original Catox
- At T-in > 430 C: better performance than original Catox

Solution:

Installation of *1st layer* of original Pt catalyst + *2nd layer* of Pt + Pd catalyst with appropriate gas hourly space velocity

IMR-MS Technology – Catox Monitoring Applications

Monitoring of Feed Gas followed by Efficient CATOX Function During Night



- **In-Process Monitoring @ Key locations**

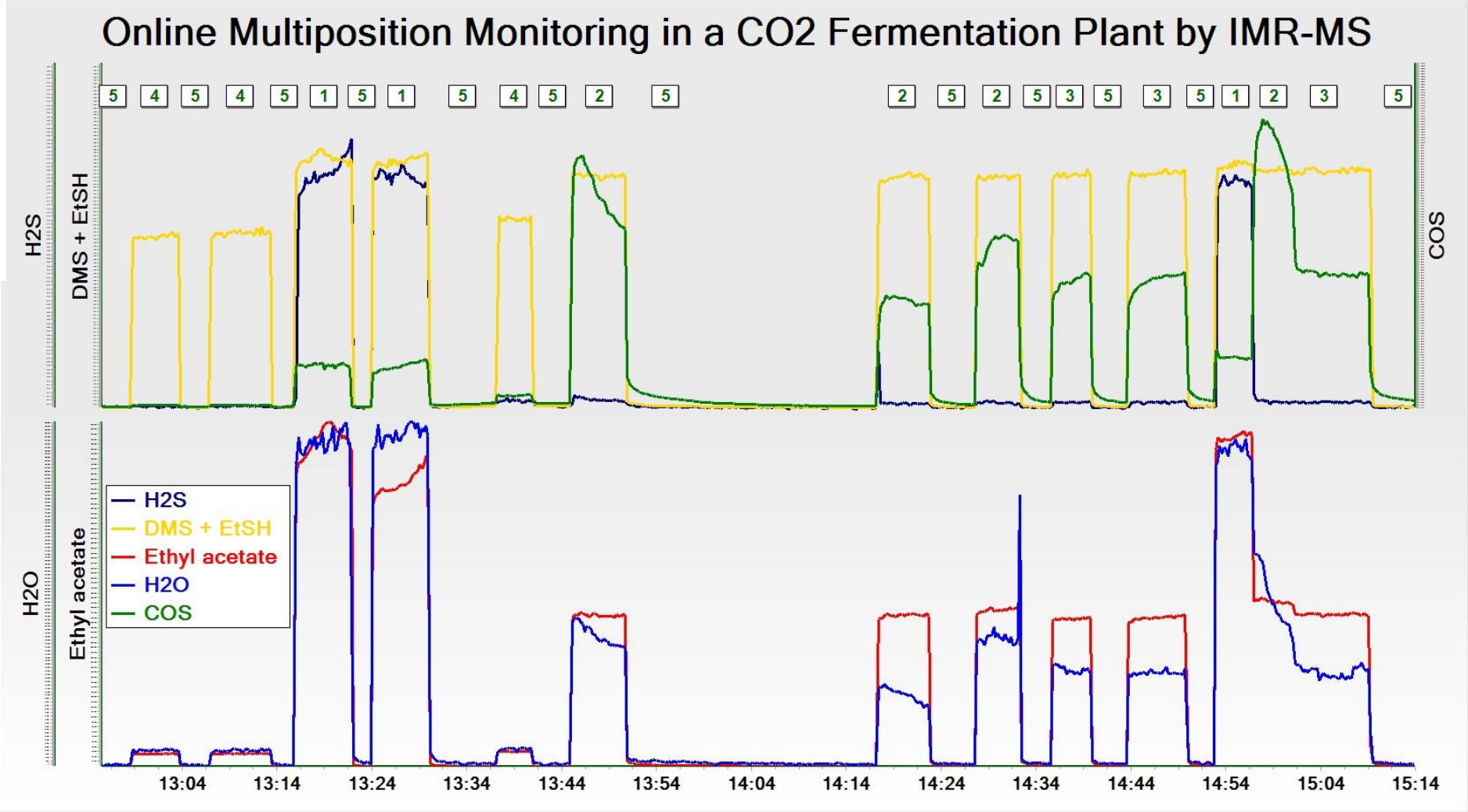
- Requirements

- *Fast respond to process & concentration changes from Feed Gas - In-Process - Final Product*
 - Rapid Sample location switching.
 - Adequate temperatures of wet gas streams
 - Regulated / equal pressures & flow rate for each sample location



IMR-MS Technology – Fermentation Plant In-Process Monitoring

- 1...Feed Gas
- 2...Post Hydrocat
- 3...Post Hydrolyzer
- 4...Post Scrubber
- 5...Product Gas to Storage



IMR-MS Technology – Natural Well Monitoring vs GC Data

Compounds	GC Result	IMR-MS Result	Difference GC vs. IMR-MS [%]
Methane [ppm]	3,100	3,140	1.3
Ethane [ppm]	1,200	1,300	8.3
Propylene [ppm]	0.20	0.23	15.0
Benzene [ppb]	18	15	-16.7
Toluene [ppb]	4,310	4,310	0.0
Hydrogen Sulfide [ppm]	0.54	0.54	0.0
Carbonyl Sulfide [ppm]	1.7	1.4	-17.6

Total Sulfur Converter

- It is important to monitor Total Sulfur Content (TSC) in both feed and final product CO₂.
- TSC is present in most CO₂ feed gas sources.
- High TSC has caused many beverage recalls.
- Unexpected increases of TSC in feed gas can cause liquid CO₂ quality upsets.
- Most on-line analyzers do not have the wide measurement range and H₂O tolerance for single instrument TSC analysis for both feed gas and final product.

ISBT Bulk Carbon Dioxide Quality Guidelines and Analytical Methods Reference 2010
(Method 13.0, recommends a TSC Limit = 0.1 ppm v/v (IMR-MS [SIS] is an acceptable method for this application)



Background & Definitions

Carbonyl Sulfide (COS) is the most important Volatile Sulfur Compound to monitor.

- COS is odorless.
- COS can be created during the CO₂ manufacturing process and is very common in feed gas sources.
- COS slowly hydrolyses into highly odiferous H₂S (rotten eggs smell) in acidic conditions such as carbonated beverages.
- High COS is historically responsible for most sulfur related odor complaints and packaged beverage recalls.



Background & Definitions

TSC Objective: ISBT Method 13.0 – Convert all forms of Volatile Sulfur Compounds (VSC) in CO₂, into a single form for measurement as Total Sulfur Content. This can be achieved by either a reductive or oxidative conversion.

Reductive TSC Reactor



(Ni Catalyzed +  T=1000 °C)



Undesired Side Reaction at > 1010 °C

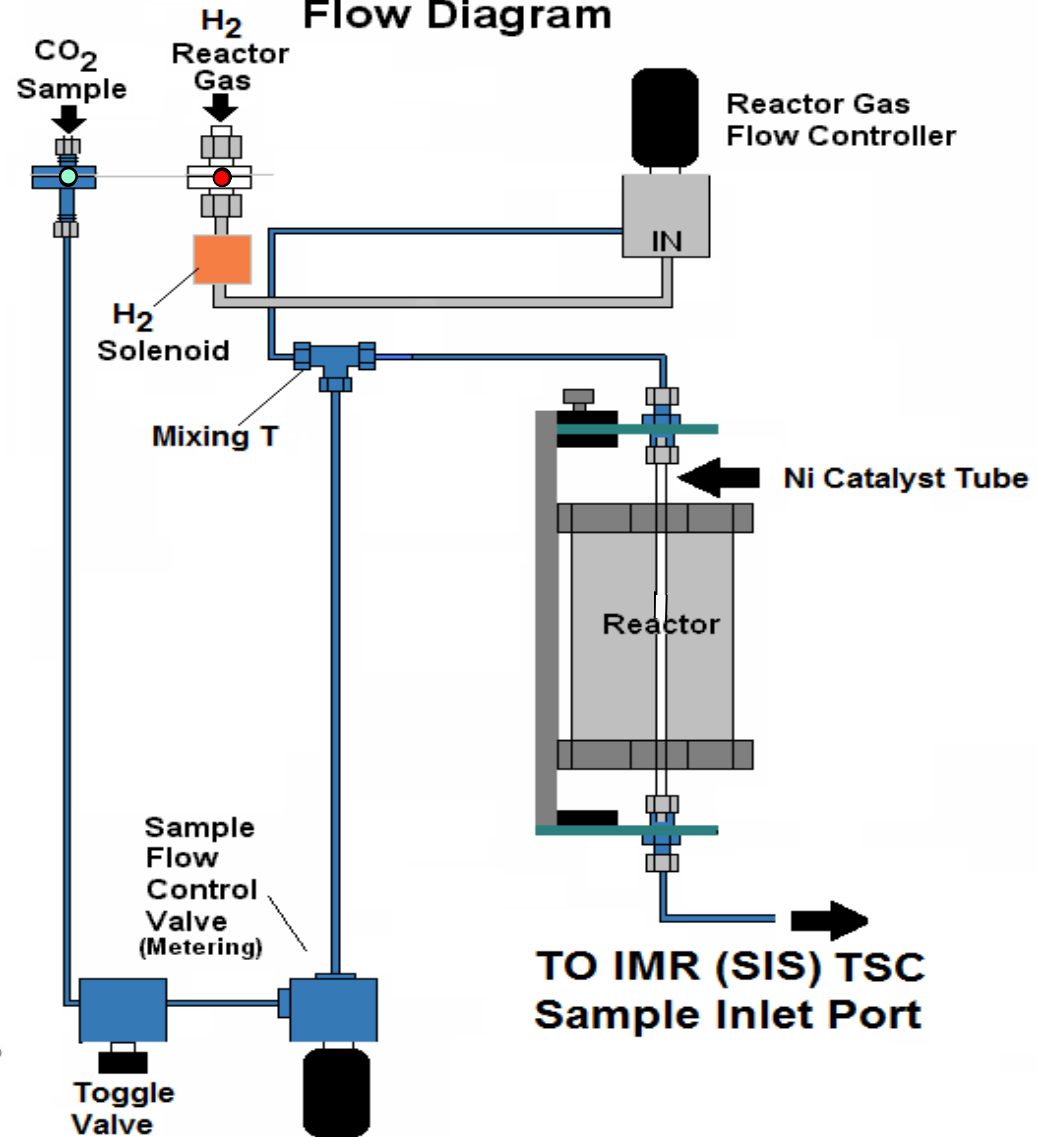


*Atomic Mass (AM) = 34 amu

TSC-SIS Hardware / Plumbing

*SulfAlert-1*TM

Flow Diagram



- Sample
- Hydrogen
- Total Sulfur

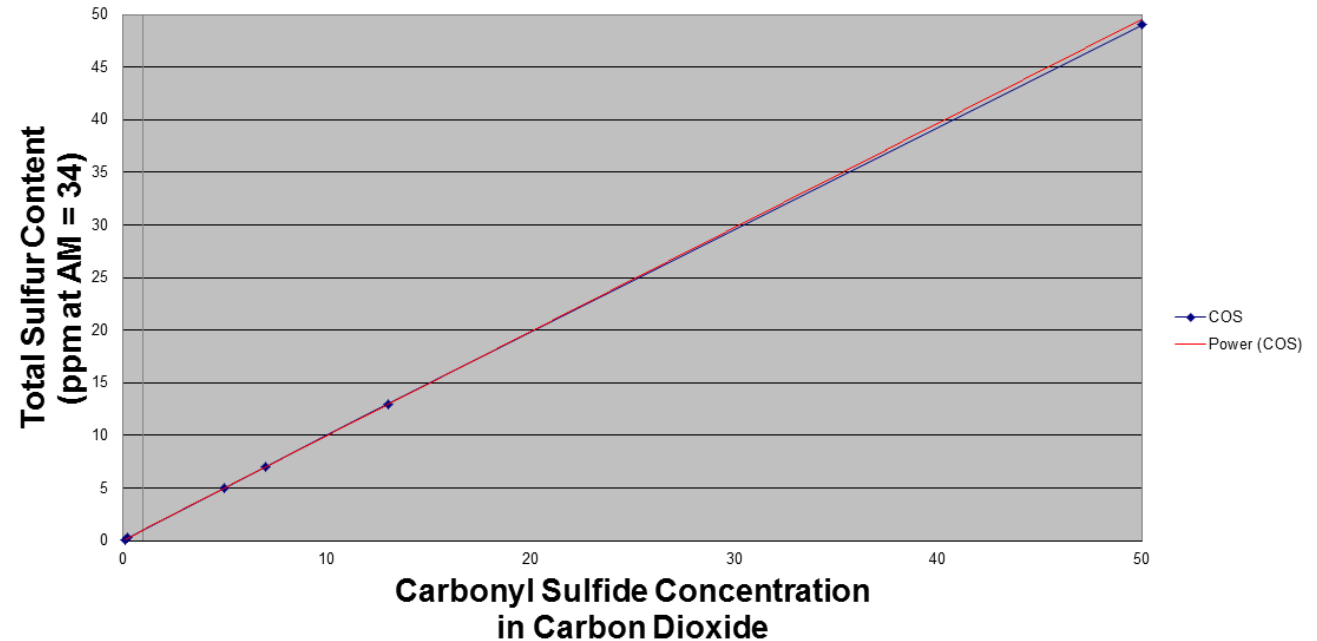
TSC Performance Data

TSC Measurement Range

- ISBT Method 13.0 recommended working range: 0.00 – 1 ppm v/v
 - Limits: 0.1 ppm v/v TSC and 1.0 ppm v/v SO₂.
- Current range is 0.00 ppm v/v to ≈50 ppm v/v*.
- Most commercial feed gas sources have TSC < 50 ppm.

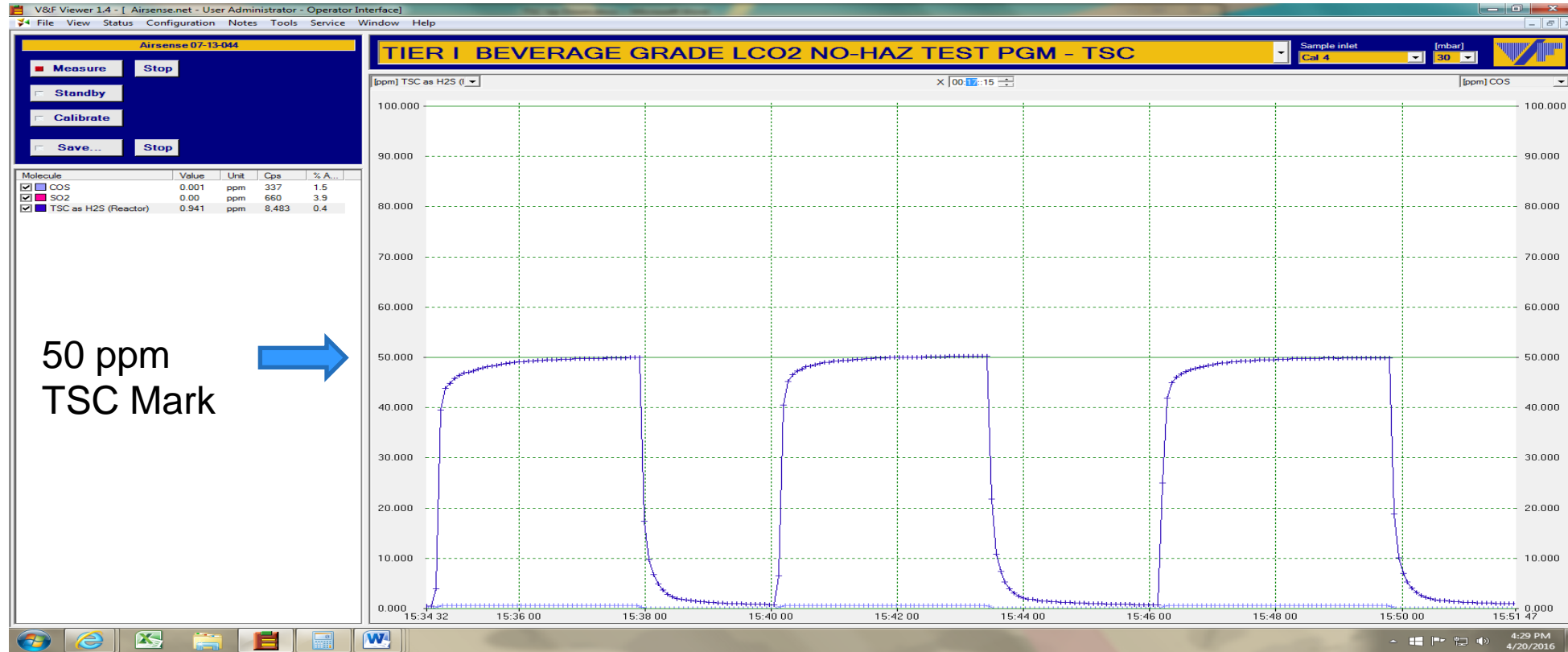
* Higher ranges not studied to date.

Conversion of COS to TSC
Using IMR-MS (SIS) and Reductive Reactor



TSC Performance Data

TSC Measurement @ 50 ppm Level



50 ppm
TSC Mark



Alternating analysis from 50 ppm COS to Zero CO₂

% COS conversion to TSC \approx 98+%

Key Sulfur Species currently being Studied at Feed Gas Impurity Levels

- Carbonyl Sulfide (COS)
- Hydrogen Sulfide (H₂S)
- Dimethyl Sulfide (DMS)
- Ethyl Mercaptan (EtSH)
- Carbon Disulfide (CS₂)
- Sulfur Dioxide (SO₂)

Advantage:

- Both TSC & speciated VSC are *simultaneously* monitored by the IMR-MS (SIS) with TSC reactor outlet.

Summary

IMR-MS (SIS) analyzers equipped with dual capillary inlets and reductive TSC converters exhibit a wide measurement range which is required for:

- Continuous CO₂ feed gas monitoring.
- Continuous in-process CO₂ composition.
- Continuous monitoring of beverage grade CO₂ for TSC and other key parameters.





V&F Analyse- und
Messtechnik GmbH



Airborne Labs International

Questions?

Thank you for your Interest!